A NEW MODE OF <u>in vitro</u> CYCLIZATION OF HUMULENE, FORMATION OF 3,6-SECOPROTOILLUDANE SKELETON

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In the hypothesis of illudoids¹ biosynthesis², formation of the key intermediate protoilludyl cation 2 has been reasonably explained by the cationic cyclization of humulene, initiated at the $\Delta^{9,10}$ double bond (Scheme I). However, to date, all attempts³ of chemical cyclization afforded products resulting from the initial electrophilic attack at $\Delta^{6,7}$ bond. We now report a new mode of humulene cyclization of the former type, which leads to the 3,6-secoprotoilludane skeleton⁴.

Oximercuration⁵ of humulene with an excess of $Hg(OAC)_2$ in a mixture of THF and H_2O (1/1) (rt, 6hr) and subsequent demercuration with NaBH₄ (rt, 2hr) gave a mixture of two tricyclic ethers 3 and 4 in 60% yield (3/4 = 3/1; determined by vpc and nmr). Separation of the ethers was performed by means of column chromatography on AgNO₃-silicagel⁶ to give pure samples. 3^7 : oil; nmr(CCl₄) $\delta 0.97$ (3H, s) 1.09 (6H, s), 1.13 (3H, s); ir(neat) $1050cm^{-1}$; mass 222 (M⁺=C₁₅H₂₆O⁺), 109 (base peak). 4^7 : oil; nmr(CCl₄) $\delta 0.95$, 1.02 (each 3H, s), 1.12 (6H, s), 5.54 (2H, s); ir(neat) 3030, 1100, 1075, $1040cm^{-1}$; mass 220 (M⁺=C₁₅H₂₄O⁺), 109 (base peak). The olefinic ether 4 was readily hydrogenated (H₂, PtO₂, EtOH) to give the saturated ether 3 quantitatively. The nmr spectrum of 5^7 , mp 108°~111°, derived from 4 (OsO_4 , Pyr, rt, 1hr; NaHSO₃, 95%) exhibited peaks at $\delta(CCl_4)$ 1.30 (6H, s,

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 $CH_3 - (-0 \times 2)$, 1.85 (1H, q, H_A, $J_{AB} = 14Hz$, $J_{AX} = 6Hz$), 3.28 (1H, d, H_Y, $J_{XY} = 4Hz$ (1H, d×q, H_X, $J_{XY} = 4Hz$, $J_{AX} = 6Hz$, $J_{BX} = 12Hz$) and indicated the presence of t tial structure A. Since the nmr spectra of 3, 4 and 5 exhibited no signal the methine group bearing ethereal oxygen atom, the ether moiety CH_3 - -0-was obtained as a partial structure for these compounds. The partial structure doubled with mechanistic considerations⁸ suggested a formula like 4 for t olefinic ether. Treatment of 4 with MCPBA (CH_2Cl_2 , rt, 100%) gave an epox mp 39° $\sim 40.5^{\circ}$, [nmr(CCl_4) $\delta 2.48$ (1H, d, J=4.5Hz), 3.07 (1H, t, J=4.5Hz)] was converted to an alcohol Z^7 (80%), mp 75.5° \sim 76.5°, nmr(CCl_4) $\delta 3.09$ (1 J=3.5Hz). The Jones oxidation of 7 afforded a ketone 8^7 (90%), mp 56.5° - nmr(CCl_4) $\delta 0.99$, 1.09, 1.27, 1.31 (each 3H, s); ir(nujol) 1735cm⁻¹; mass ($M^+=C_{15}H_{24}O_2^+$).

Elucidation of the structures 3 and 4 was achieved by X-ray crystal lographic analysis of the keto-ether 8: $P2_1/c$, a=9.558(5), b=6.195(3), c= (10)Å, β =109.48(5)°, Z=4. The structure was solved by the direct method a fined by the block-diagonal-matrix least-squares method using 2109 indepereflections (Cu Ka, $\theta \leq 70^\circ$). The final R value including hydrogen atoms was The perspective view of 8 thus obtained is shown in Fig.2 and the final a coordinates are given in Table 1.

The ethers 3 and 4 here obtained correspond to ether analogs of the thetical biosynthetic intermediate 1 and its dehydro compound respectivel Therefore it is hoped that these ethers could be transformed to illudoids in vitro processes. Studies along this line are now in progress.

Table I. The final atomic coordinates

Atom	x/a	y/b	z/c	Atom	x/a	y/b	z/c
0(1)	-0.1445	-0.0311	0.10061	C(8)	-0.0265	0.4022	0.08159
0(2)	0.0676	0.3923	0.18856	C(9)	0.1317	0.3769	0.08029
C(1)	0.3929	0.2976	0.13255	C(10)	0.1709	0.1667	0.05583
C(2)	0.2545	0.4032	0.14078	C(11)	0.3417	0.1746	0.07261
C(3)	0.2128	0.3117	0.19302	C(12)	0.3150	0.4087	0.25041
C(4)	0.2134	0.0611	0.19801	C(13)	-0.1934	0.4008	0.14295
C(5)	0.0601	-0.0293	0.19134	C(14)	0.3838	0.3041	0.02623
C(6)	-0.0530	0.0702	0.13893	C(15)	0.4096	-0.0497	0.07773
C(7)	-0.0485	0.3160	0.13738				

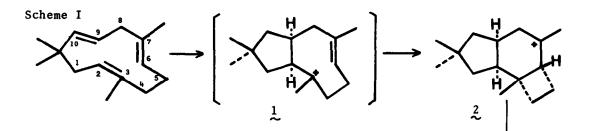
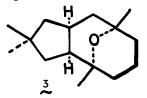
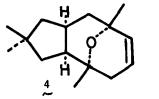
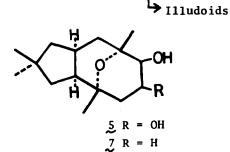
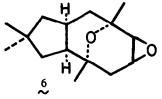


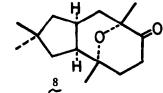
Figure I

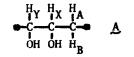






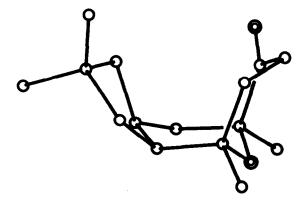






= a quarternary carbon atom

Figure II



References and notes

- Naming: Y.Ohfune, H.Shirahama and T.Matsumoto, Tetrahedron Lett., 4377 (1975).
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 e) An effective preparation of 9,10-epoxy humulene was very recently developed; A.Satter, J.Forrester, M.Moir, J.S.Roberts and W.Parker, Tetrahedron Lett., 1403 (1976).
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- 5. H.C.Brown and P.Geoghegan, Jr., J.Amer.Chem.Soc., 89, 1523 (1967).
- 6. A.C.Gupta and Sukh Dev, J.Chromatography, 12, 189 (1963).
- 7. Satisfactory elementary analytical data were obtained.
- Although the detailed mechanism for their formation is not clear, the following process may be suggested.

humulene

